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Pre-lab Assignment/Questions *Note- this pre-lab must be finished before you come to lab. 1. Calculate the molar mass of citric acid C 6 H 8 O 7. 2. Calculate the volume of 2.5 M NaOH needed to make 200 mL of 0.04 M NaOH. 3. The balanced equation for KHP and NaOH is given in the introduction to this lab.

Lab 7 Worksheet | College Chemistry 1 Labs
CH4(g) +O2(g) → CO2(g) +2H3(g)O ΔH= −890.4 kJ CH 4 (g) + O 2 (g) → CO 2 (g) + 2H 3 (g) O Δ H = − 890.4 kJ. The negative sign serves to reinforce the fact that heat is produced rather than absorbed or that this is an exothermic reaction. We can use stoichiometry to calculate the enthalpy of any amount of a reactant.

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