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Modern Chemistry Chapter 10 Section 4 Review Answers

CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

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Chapter 10 Review States of Matter (Section 2) Molecules in liquids are very mobile. This mobility allows a liquid to take the shape of its container. In liquids molecules are adjacent with molecules, allowing intermolecular forces to have a greater effect than they do in gases.

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Name SECTION 2 Date CHAPTER 11 REVIEW Gases Class SHORT ANSWER Answer the following questions in the space provided. 1. State whether the pressure of a fixed mass of gas will increase, decrease, or stay the same in the

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