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2 Chemistry: Matter and Change Solving Problems: A Chemistry Handbook SOLVING PROBLEMS: CHAPTER 1 A CHEMISTRY HANDBOOK Matter is made up of particles, called atoms, that are so small they cannot be seen with an ordinary light microscope. The struc-ture, composition, and behavior of all matter can be explained by atoms and the changes they undergo.

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A change in physical form of a substance or physical properties of a substance like color, melting point, state, size, shape, and arrangement of molecules is known as physical change. A change in chemical properties of a substance like composition, reactivity with water, corrosiveness, and pH is known as chemical change.

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Solutions Manual Chemistry: Matter and Change • Chapter 11 211 CHAPTER 11 SOLUTIONS MANUAL Section 11.1 Assessment page 372 5. Compare the mass of the reactants and the mass of the products in a chemical reaction, and explain how these masses are related. The coefficients in the balanced equation indicate the molar relationship between each ...

### StoichiometryStoichiometry

iv Chemistry: Matter and Change Study Guide for Content Mastery This Study Guide for Content Masteryfor Chemistry: Matter and Change will help you learn more easily from your textbook. Each textbook chapter has six study guide pages of questions and exercises for you to complete as you read the text.

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