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for Chemistry chapter
3 (Atomic Structure) In
this topic, Student
should be able to : a)
Identify and describe
the proton, neutron
and electron in terms
of their relative
charges and relative
masses. b) Discuss the
behaviour of beams of
protons, neutrons and
electrons in electric
fields.

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Chapter Test A

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table below to answer

the questions in this

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space provided, write

the letter of the term

or phrase that best

completes each

statement or best

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Answers
of an atom in a chemical compound to attract electrons from another atom in the compound is called _____. 17. The energy required to remove one electron from an atom is called its _____. 18.

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Chemistry 3 Chapter
Test Chapter Test B,
continued 16. The
measure of the ability
of an atom in a
chemical compound to
attract electrons from
another atom in the
compound is called
_____.

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Answer Key

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19 Questions Show
Answers. Question 1

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1 Chemistry: A
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Molecular Approach, 3e
(Tro) Chapter 3

Molecules, Compounds,
and Chemical

Equations Multiple

Choice Questions 1) An

ionic bond is best

described as A) the

sharing of electrons. B)

the transfer of

electrons from one

atom to another. C) the

attraction that holds

the atoms together in a

polyatomic ion. D) the

attraction between 2

nonmetal atoms.

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the same as in chapter
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3 due Monday
February 25 at the
Beginning of Class You
must show all work to

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Key More Exam 2

Practice Quiz6 Quiz6

Key Q7a&bKeys Quiz 8

Key: Quiz 9 Due Wed.

April 17th

CHM151 Quizzes and Exams

MDCAT Chemistry

Chapter 3 MCQ Test

With Answer for

Chemistry chapter 3

(Atomic Structure) In

this topic, Student

should be able to : a)

Identify and describe

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Answers
the proton, neutron and electron in terms of their relative charges and relative masses. b) Discuss the behaviour of beams of protons, neutrons and electrons in electric fields.

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Form 4 1. Name : _____

Class : _____ TOPICAL

TEST CHAPTER 3 1.

Write down the
chemical equation of
the following reactions

a. Sodium reacts with
oxygen to produce a
white metallic oxide.

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ecf8427e.

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