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Terms in this set (15)

the amount of products

that form when a

reaction is carried out

in the laboratory.

actual yield. the

calculation of

quantities in chemical

equation.

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Study 15 Terms | 9

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PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. 1.

Knowing the mole ratio of a

reactant and product in a

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chemical reaction

would

allow you to determine

Name Class Date Assessment)

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Introduction to

Stoichiometry

Composition

Stoichiometry - deals

with mass relationships

of elements in

compounds Reaction

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Involves mass relationships between reactants and products in a chemical reaction

I. Reaction

Stoichiometry

Problems A. Four

problem Types, One

Common Solution

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Chapter 9

STOICHIOMETRY. What

is Sto-Key-Om-It-Tree?

Stoichiometry is

derived from the Greek

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words stoicheion

“element” and metron

“measure”.

Stoichiometry is defined as the branch of chemistry that deals with the mass relationships between elements in a compound

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PC\|MAC

The reaction stoichiometry problems in this chapter can be

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classified ... operate
Instruments to test

municipal water
supplies for pH levels
and the levels of lead.

Chemical technicians

... . Chapter 9 DO NOT
EDIT--Changes must be
made through "File
info" CO +) ...

CorrectionKey=NL-A

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be made ...**

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Multiple Choice Identify

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Test Answers

the choice that best completes the statement or answers the question. ****You need to know all the rules for naming ionic, molecular and acidic compounds. ****You will have questions over ch 1,2,4 and 25 as well. The majority of your test is naming though.

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reactants and
products. CHAPTER 9
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Chapter 9 Review

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Answers

Stoichiometry SHORT ANSWER Answer the following questions in the space provided. 1. b The coefficients in a chemical equation represent the (a) masses in grams of all reactants and products. (h) relative number of moles of reactants and products. (c) number of atoms of each element

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ineach compound a
reaction. Answers

Date. FCHAPJ

REV[EW.

CHAPTER 9 REVIEW

Stoichiometry SECTION

1 SHORT ANSWER

Answer the following questions in the space provided. 1. b The coefficients in a chemical equation represent the (a) masses in grams of all reactants and products. (b) relative

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Test Answers

number of moles of reactants and products. (c) number of atoms of each element in each compound in a reaction.

mc06se cFMsr i-vi

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7 9 2 6 3 5 1 8 _____ 13

_____ 12 _____ 14 _____ 15 _____

10 _____ 11 DIRECTIONS:

Write on the line at the right of each statement the letter preceding the word or expression that best completes the statement. 1.

Which of the following would be investigated in reaction stoichiometry?

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Problems 12 Section

121 1 10A Answers

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& Problem Set • Read

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can skip all of section

9.3) • Lab 8:

Quantitative Analysis

• Regents Tables :

Table T : Important

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Equations • Warm-ups and problems will be collected before you take the test. Answer all problems in the space provided.

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g of NH_3 is actually
obtained in the lab
during the following
reaction: $4\text{NH}_3 + 5\text{O}_2$
 $\rightarrow 4\text{NO} + 6\text{H}_2\text{O}$ How
many grams of NO are
formed if 6.

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