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With Answer for

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(Groups) In this

topic, Student should

be able to : Discuss the

variation in the

physical properties of

elements belonging to

period 2 and 3 and to

describe and explain

the periodicity in the

following physical

properties of elements.

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communication,  
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importance and

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163 CHAPTER 10

SOLUTIONS MANUAL 3.

Apply How many atoms  
of hydrogen-1 are in a  
1.007-g sample? Recall  
that 1.007 amu is the  
mass of one atom of  
hydrogen-1. Round  
your answer to two

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significant digits. 1.007  
g 1 H-1 atom  $\underline{\quad}$  1.672  
10 24 g a. 6.023 10<sup>23</sup>  
H-1 atoms 4.

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Example 10.1 to  
answer the question. •

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The atomic number ( $Z$ ) = the number of protons. • The mass number ( $A$ ) = the number of protons + the number of neutrons. • Isotopes are written with the mass number to the upper left of the element symbol and the

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Questions.  $1.09 \times 10^4$

kg P g P mol P mol P

40 10 ...  
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## **Chapter 10 Chemical Calculations and Equations**

CHAPTER 10 REVIEW

States of Matter

SECTION 3 SHORT

ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left.

b ionic crystal (a) has

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mobile electrons in the  
crystal c covalent  
molecular crystal (b) is  
hard, brittle, and  
nonconducting a  
metallic crystal (c)  
typically has the lowest  
melting point of the  
four

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Solutions, Important  
Questions for 10.

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Sample Questions ...**

Review of Chemistry of  
Matter (Chapter 2, & 3  
Test Review) ANSWERS  
Directions: Review and  
check all your answer  
with this answer sheet  
to get ready for the

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test on Chemistry of  
Matter (Chapter 2 & 3)  
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Complete: Yes:

09/26/19: 50 students

verified as accurate:

Answer Tolerances:

Yes: 09/29/19: 31

students verified as

accurate: Biology: Yes:

10/06/19: 76 students

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verified as accurate:

Biology Chapter 1: Yes:

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verified as accurate:

Biology Chapter 11:

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