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volcano in Hawaii is the
most active volcano in
the world. It has been
erupting for centuries.
The hot lava oozes and
flows, scorching every-
thing in its path,
occasionally
overrunning nearby
houses. When the lava
cools, it solidifies into
rock. The properties of

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Particles in liquids are attracted to each other. These attractions keep the particles in the liquid close together, which is why liquid has a definite volume. These attractions reduce the amount of space between the particles in a liquid.

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of liquids. STUDY.
PLAY. what factors
determine the physical
properties of a liquid.
the interaction
between the motions
of particles in a liquid
and attractions among
them. fluid. anything
that has the ability to
flow --allows to fill any
container liquids AND
gas.

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Q. The attraction that
causes water and other
liquids to form drops

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on thin films is called adsorption. This is also water's ability to be attracted to other water molecules.

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(R.G. Drage) A model
for liquids Homework
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there is a balance
between the motion of

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particles and the attraction of particles for each other in a liquid these attractions keep the particles close together, which is why

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One of the key defining properties of liquids is their ability to flow. Beyond this feature, though, the behaviors of different liquids span

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a broad range. Some liquids flow relatively easily, like water or oil, while others, like honey or molasses, flow quite slowly. Some are slippery, and some are sticky.

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measure of the force
exerted by a gas above
a liquid.

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What is In a system at
constant vapor
pressure , a dynamic
equilibrium exists

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between the vapor and the liquid. the system is in equilibrium because the rate of evaporation of liquid equals the rate of condensation of vapor.

300. give an example of boiling point . 300.

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Liquids have definite volume, but not a definite shape. Liquids are said to have low

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compressibility; in other words, it's hard to pack liquid particles closer together. Compared to gases, there is...

The Kinetic Molecular Theory: Properties of Solids and Liquids

Liquids 13.2 The Nature of Liquids
Gases Particles are attracted to each other
Has kinetic energy Can take shape of container

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Has kinetic energy Can
take shape of container
-Liquids evaporate
faster when heated
because heating the
liquid increases the
average kinetic energy
of

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Home → Characteristics and Nature of Liquids. The liquid state is the intermediate state between gaseous state and solid state. In liquid state particles

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are close together in comparison to those in gases but they are relatively loose in comparison to solids. Similarly, the interparticle forces are stronger than those in gases but relatively weaker than solids.

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